# Lab-Scale Development of a Solid Sorbent for CO<sub>2</sub> Capture Process for Coal-Fired Power Plants



### DE-FE0026432

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Mustapha Soukri, John Thompson, Ignacio Luz, Jak Tanthana, Marty Lail and Kelly Amato August 24, 2017

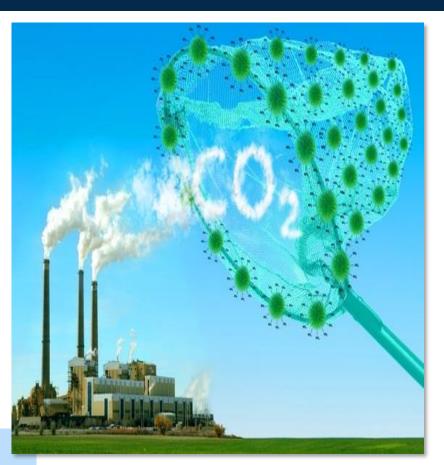
# 2017 NETL CO<sub>2</sub> Capture Technology Project Review Meeting August 21-25, 2017 - Pittsburgh



### **DOE Program Manager: Steve Mascaro**

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# Lab-Scale Development of a Solid Sorbent for CO<sub>2</sub> Capture Process for Coal-Fired Power Plants



### **Project Details – DE-FE0026432**

- Funding: \$1,989,415
  - *♦ \$1,591,532 DOE*
  - \* \$ 397,883 Cost Share State of North Carolina

### Period: October 2015 – March 2018

### Goals/Objective:

- Develop novel 3<sup>rd</sup> generation fluidizable solid sorbents for RTI's sorbent-based CO<sub>2</sub> capture process:
  - Fluidizable, hybrid-metal organic frameworks
  - \* Fluidizable hybrid-phosphorus dendrimers

# **Project Outline**

- Design and synthesize two novel fluidizable CO<sub>2</sub> adsorbents.
- Demonstrate the superior performance of these advanced CO<sub>2</sub> solid sorbents at the lab scale.

- Evaluate the impact of flue gas contaminants such as SO<sub>x</sub>, NO<sub>x</sub>, O<sub>2</sub>, and H<sub>2</sub>O on these advanced solids sorbents
- Conduct a high level techno-economic analysis.

# **Project Structure – Budget Period 1**

**Objective:** Develop several novel hybrid solids sorbent as well as packed-bed reactor testing.

*Timeframe*: 10/1/15 to 3/31/17 (18 months)

Task	Description	Objectives / Activities
1	Project Management and Planning	• Coordinate, manage and plan project activities that will include, monitoring and controlling of project scope, technical, budgetary and scheduling activities, project and task planning, asset management, cost tracking, progress reporting and updating Project Management Plan document appropriately.
2	Hybrid MOF-based CO <sub>2</sub> adsorbents	<ul> <li>2.1 – hybrid MOF-based sorbents synthesis and development.</li> <li>2.2 – Hybrid MOF-based sorbents evaluation and optimization.</li> <li>2.3 – Molecular Modeling of Hybrid MOF-based sorbents.</li> </ul>
3	Hybrid <i>P</i> -Dendrimer-based sorbents	<ul> <li>3.1 – hybrid <i>P</i>-Dendrimer-based sorbents synthesis and development.</li> <li>3.2 – Hybrid <i>P</i>-Dendrimer-based sorbents evaluation and optimization.</li> <li>3.3 – Molecular Modeling of Hybrid <i>P</i>-Dendrimer-based sorbents.</li> </ul>
4	Long-term Performance Testing and Technical Merit Comparison	<ul> <li>4.1 – Multi-cycle performance testing of most promising <i>P</i>-Dendrimer-based and MOF-doped sorbents.</li> <li>4.2 – Preliminary sorbent production cost review</li> </ul>

# **Budget Period 1 Milestones**

		Task	Description	Date		
00	A	1	Update Project Management Plan	10/31/15		
eriod	В	1	Complete Kick-off meeting			
et P	С	2.2	Selection of the first 3 optimized high-potential hybrid MOF solid sorbents for $\rm CO_2$ capture.	12/31/2016		
Budget	D	D 3.3 Selection of the first 3 optimized high-potential hybrid <i>P</i> -dendrimer solid sorbents for CO <sub>2</sub> capture.		12/31/2016		
â	E	4.1	Further selection of most-promising hybrid solid sorbent based on long-term performance criteria and cost review.	03/31/2017		

# **Budget Period 1 Milestones Completed**

Milestone Description	Budget Period	Planned Completion Date	Actual Completion Date	Verification Method	Comments
A. Updated Project Management Plan	1	10/31/2015	09/23/2015	Project Management Plan file submitted	<b>Complete.</b> PMP was updated and approved by DOE/NETL. File was provided to Project Officer and is used as an ACTIVE PMP.
B. Kick-off Meeting	1	12/17/2015	12/17/2015	Presentation file submitted	<b>Complete.</b> Presentation given at on-site kick-off meeting on 12/17/2015. Presentation provided for public release to Project Officer on 01/13/2016.
<b>C</b> . Selection of the first 3 optimized high-potential hybrid MOF solid sorbents for CO <sub>2</sub> capture.	1	12/31/2016	11/30/2016	Quarterly Report #5	<b>Complete.</b> The project team has produced numerous hybrid MOF sorbents. Several high potential hybrid MOF candidates have exhibited very good $CO_2$ capture of $\ge 12$ wt% and good stability. RTI has selected three hybrid MOF candidates to move forward.
<b>D</b> . Selection of the first 3 optimized high-potential hybrid P-dendrimer solid sorbents for CO <sub>2</sub> capture.	1	12/31/2016	1/31/2017	Quarterly Report #5	<b>Complete.</b> The project team has produced over 125 hybrid <i>P</i> -dendrimer sorbents. Several high potential <i>P</i> -dendrimer candidates have exhibited good physical and performance characteristics, including $CO_2$ loading capacity $\ge 12$ wt.%. RTI has selected three hybrid <i>P</i> -Dendrimer sorbent candidates to move forward, including one in particular selected for formulation into a fluidizable form.
E. Further selection of most- promising hybrid solid sorbent based on long-term performance criteria and cost review.	1	03/31/2017	3/31/2017	Quarterly Report #6	<b>Complete.</b> Highly promising MOF and <i>P</i> -dendrimer sorbent candidates have undergone multi-cycle packed-bed reactor testing. All six highly promising candidates have been tested for at least 250 adsorption/desorption cycles. All sorbent candidates exhibit desirable performance and thermal stability. The P-Dendrimer-based sorbent candidates in particular exhibit little to no degradation in multi-cycle tests.

# **Project Structure – Budget Period 2**

**Objective:** Scale-up and testing of preferred sorbent as well as preliminary techno-economic analysis

*Timeframe*: 04/01/17 to 3/31/18 (12 months)

Task	Description	Description Objectives / Activities		
1	Project Management and Planning	Continuation of BP1 project management and planning		
5	Scale-up and Testing of Selected Candidate	<ul> <li>5.1 – Scale up production of selected sorbent in fluidizable form.</li> <li>5.2 – Performance testing in lab-scale fluidized-bed reactor system.</li> <li>5.3 – Contaminant impact testing in packed-bed reactor.</li> <li>5.4 – Preliminary review of process requirements relative to conventional equipment.</li> <li>5.5 – Optimization of selected candidate and kilogram-scale production.</li> </ul>		
6	Preliminary Techno-Economic Analysis	<ul> <li>6.1 – Preliminary process design.</li> <li>6.2 – Preliminary economic evaluation.</li> </ul>		

# **Project Milestones**

riod 2		Task	Description	Date
get Peri	F	5.1	Successful scale-up of preferred hybrid sorbent in fluidized form and experimental data from lab-scale FMBR prototype capable of achieving 90% $CO_2$ capture from simulated flue gas.	9/30/17
Bud	G	6.2	Complete technical and economical evaluation.	3/31/18

# Hybrid MOF-Based CO<sub>2</sub> Adsorbents

# Hybrid MOF-Based CO<sub>2</sub> Adsorbents

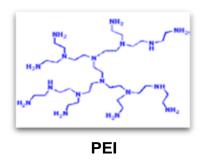


Silica

- Attrition resistance
- Fluidizable
- Low cost
- Acceptable density



- Exceptionally high surface areas
- Tunable pore sizes
- Commercially available linker



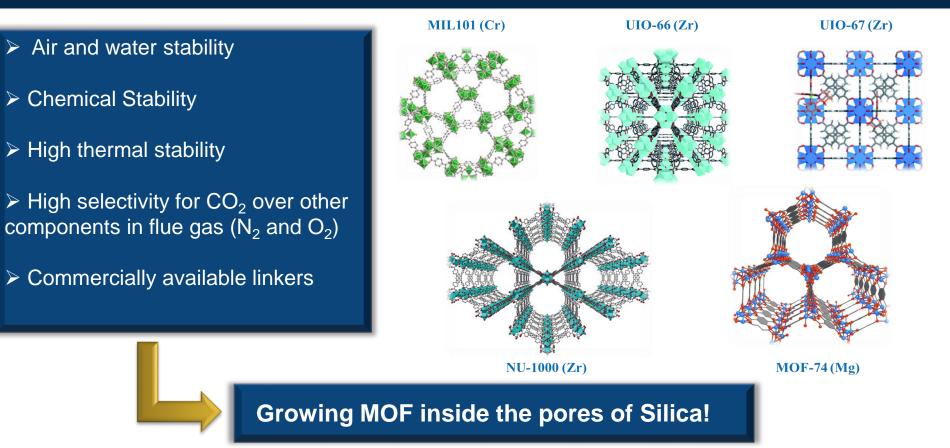
- High amine content
- High CO<sub>2</sub> affinity
- Relatively low cost materials



Table 2. Reported CO <sub>2</sub> Capture Performance Results			
Sample Description CO <sub>2</sub> Capacity (wt%)			
MOF	21.4ª		
MOF-amine	~20		
Fluidizable silica (FS)	0		
FS-PEI	4.8		
MOF-silica	0		
MOF-silica-amine	9.3		
<sup>a</sup> Final report award # DE-FC26-07NT43	092		

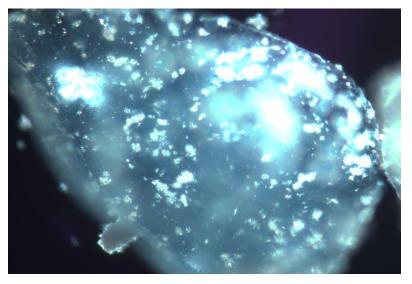
Silica + MOF + PEI

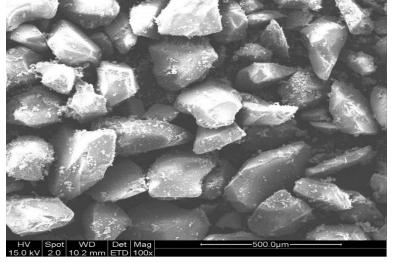
# MOFs Selected for Evaluation as Hybrid MOF-Based CO<sub>2</sub> Adsorbents



# Solvo-Thermal Synthesis of MOF-Silica Hybrid

### The State-of-Art Solvo-thermal Synthesis of MOF-Silica Hybrid is non-selective!





### **Confocal microscope picture**

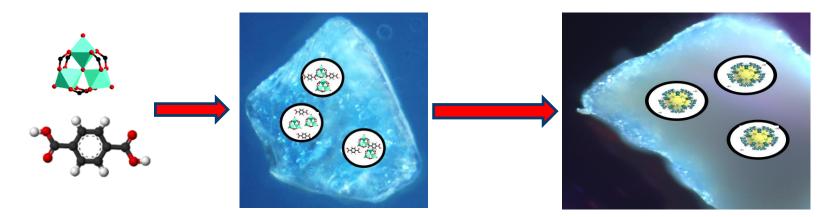
### **SEM picture**

Is the current solvo-thermal method the best approach for the MOF-Silica hybrid synthesis?

- Not utilizing the internal pores of the silica
- Poor interaction of MOF with Silica → Low yields
- Low attrition-resistance

# **New Approach for MOF-Silica Hybrid Preparation**

### **Our new approach: Solid State Synthesis**



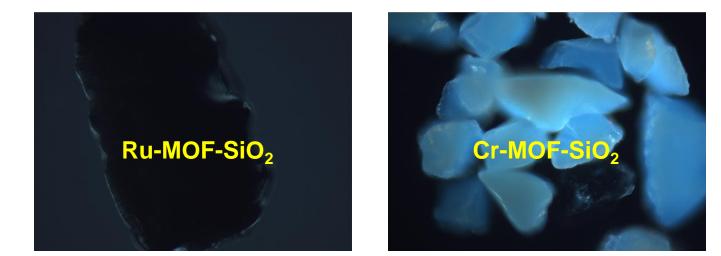
New approach allowed the project to meet the fist goal of the MOF-Silica hybrid Synthesis *via* sequential incipient wetness-impregnation/evacuation

Full characterization using the most well known technics such as: Confocal Microscope, SEM, FIB-FESEM, TEM, FTIR, XRD, XRF, N<sub>2</sub> isotherms, TGA, Particle size distribution, Jet-Cup attrition index

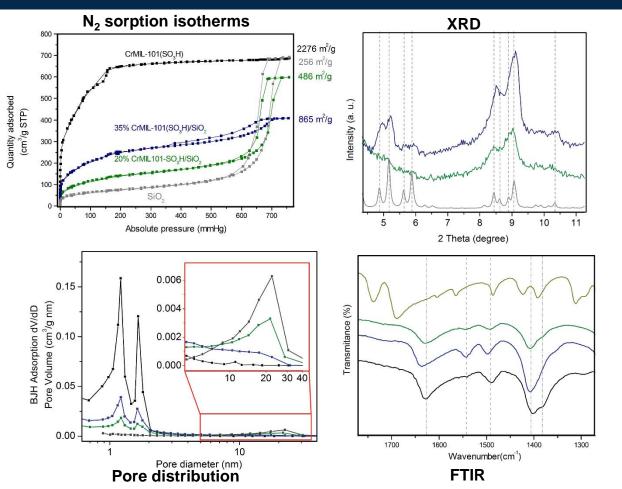
# **Confocal Microscope for the New MOF-Silica Hybrids**

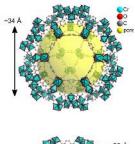


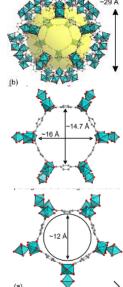
100 µm



# Full characterization for (Cr)MIL-101(SO<sub>3</sub>H)/SiO<sub>2</sub>







# **Confining Metal-Organic Framework Nanocrystals Within Mesoporous Materials: A General Approach via 'Solid-State' Synthesis.**



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#### Article

#### Confining Metal-Organic Framework Nanocrystals Within Mesoporous Materials: A General Approach via 'Solid-State' Synthesis.

Ignacio Luz, Mustapha Soukri, and Marty Lail Chem. Mater., Just Accepted Manuscript • Publication Date (Web): 21 Aug 2017 Downloaded from http://pubs.acs.org on August 21, 2017

#### Just Accepted

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#### SOLID-STATE CRYSTALLIZATION OF METAL ORGANIC FRAMEWORKS WITHIN MESOPOROUS MATERIALS METHODS AND HYBRID MATERIALS THEREOF

#### CROSS REFERENCE TO RELATED APPLICATIONS

[0010] This application claims the benefit of 62/373,047 filed August 10, 2016, Ignacio Luz, Atty. Dkt. No. 474774US which is hereby incorporated by reference in its entirety.

#### 1. FIELD

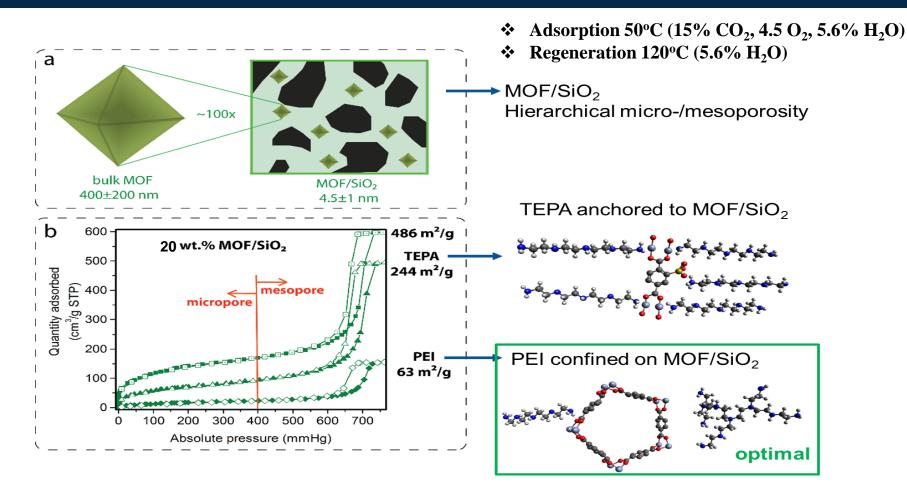
[0011] The present disclosure relates to a general method for the solid-state crystallization of metal organic frameworks (MOFs) within the pore spaces of mesoporous materials (MPMs) in the absence of solvent. Additionally, the present disclosure relates to hybrid metal organic framework (MOF) and mesoporous material (MPM) hybrid materials (MOF/MPM) generated therefrom.

2. BACKGROUND

#### 2.1. Introduction

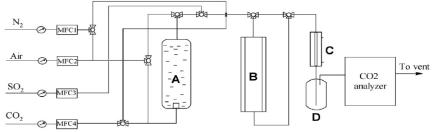
[0012] The "background" description provided herein is for the purpose of generally presenting the context of the disclosure. Work of the presently named inventors, to the extent it is described in this background section, as well as aspects of the description which may not otherwise qualify as prior art at the time of filing, are neither expressly or impliedly admitted as prior art

# **Polyamine-Containing MOF/SiO<sub>2</sub> Fluidized Sorbents**



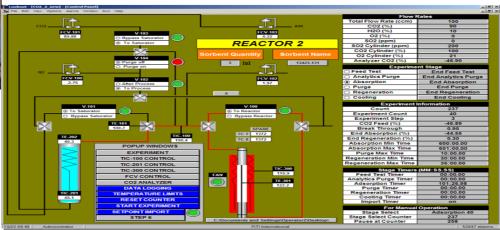
# **Packed-bed reactor (PBR)**



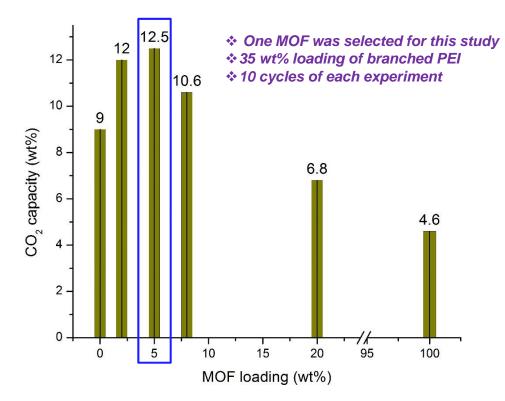


- Fully-automated operation and data analysis; multi-cycle absorption-regeneration
- Rapid sorbent screening experiments
- > Measure dynamic  $CO_2$  loading & rate
- > Test long-term effect of contaminants
- <u>2 gram of hybrid CO<sub>2</sub> solid sorbent</u> is used for testing

<u>Adsorption</u>:  $CO_2=15 \text{ vol\%}$ ,  $O_2 = 4.5 \text{ vol\%}$ ,  $H_2O= 5.65 \text{ vol\%}$ in balance with  $N_2$  at 50 °C <u>Regeneration</u>:  $H_2O = 5.65 \text{ vol\%}$  in balance with  $N_2$  at 120 °C



# Effect of MOF Loading on CO<sub>2</sub> Capture



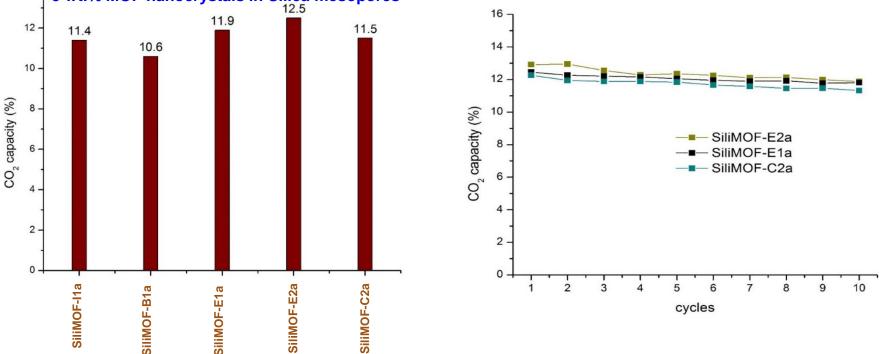
- 5 wt.% MOF nanocrystals in Silica mesopores lead to an improvement of CO<sub>2</sub> capture capacity (12.5 wt.% CO<sub>2</sub>)
- Higher MOF loadings undergoes partial inhibition of amine efficiency to capture CO<sub>2</sub>
- This effect has been observed for 10 different hybrid-MOF/SiO<sub>2</sub> exhibiting different features, such as surface area, functional groups, open metal sites and pore size.

#### Simulated flue gas composition:

 $CO_2 = 15$  vol%,  $O_2 = 4.5$  vol%, and water = 5.65 vol% balanced with  $N_2$ 

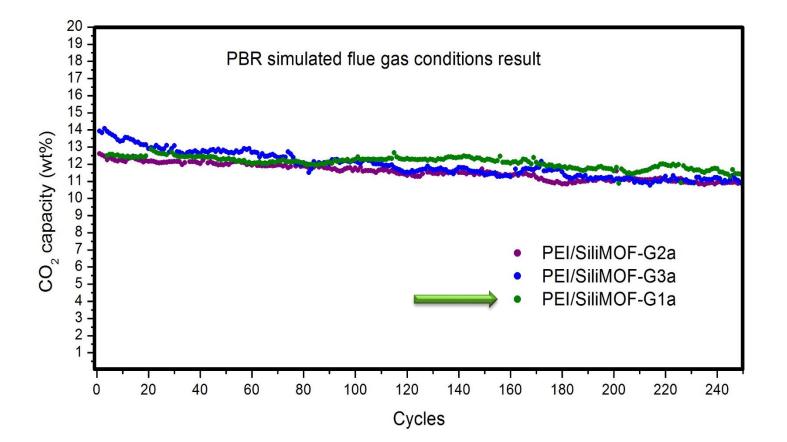
## **Effect of MOFs Structure on CO<sub>2</sub> Capture**

5 wt.% MOF nanocrystals in Silica mesopores

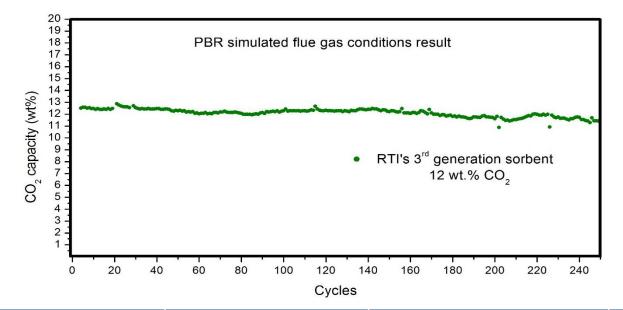


### Simulated flue gas composition: $CO_2 = 15 \text{ vol}\%$ , $O_2 = 4.5 \text{ vol}\%$ , and water = 5.65 vol% balanced with $N_2$

# **Multi-Cycles Testing for 3-Selected Sorbents Candidates**



# **Multi-Cycles Testing for Selected Sorbent Candidate**



Candidate	Packed bed reactor-CO <sub>2</sub> Capacity (wt%)		Chemical Makeup		Thermal
	Capacity	Cycles	Pre Analysis	Post Analysis	Stability
PEI/HyperMOFG1a	12.5%	250	N: 13.38% C: 22.36% H: 4.84%	N: 12.05% C: 20.57% H: 4.35%	up to 150 ℃

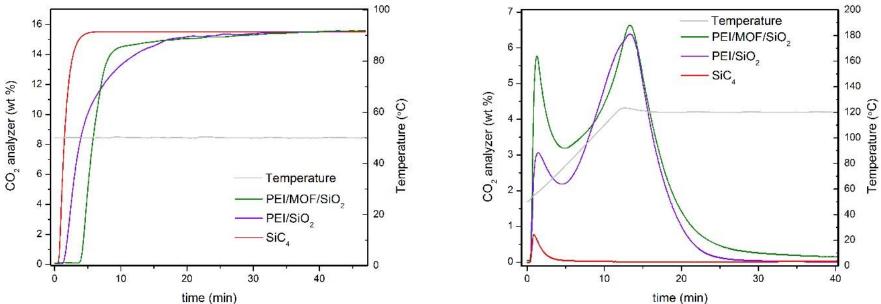
# **CO<sub>2</sub>** Capture in Packed-Bed Reactor

### **Adsorption**

15 vol% CO<sub>2</sub>, 4.5 vol% O<sub>2</sub>, 5.6 vol% H<sub>2</sub>O, N<sub>2</sub> (50 °C)

### Regeneration

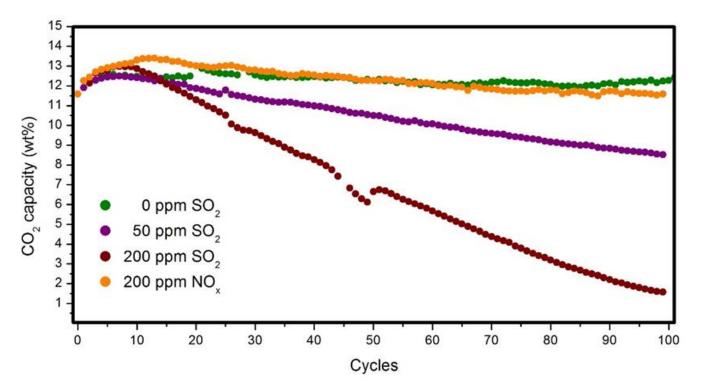
 $H_2O = 5.6 \text{ vol}\% + N_2 (120 \text{ °C})$ 



**Unprecedented dual physi-chemisorption** 

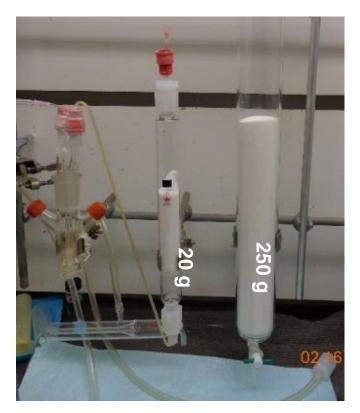
### SO<sub>2</sub> and NOx Contaminants Testing for Hybrid *MOF*-Based Sorbent

- Completed testing of 50 and 200 ppm SO<sub>2</sub> levels
- 100 cycles testing of NOx (200 ppm) in simulated flue gas streams



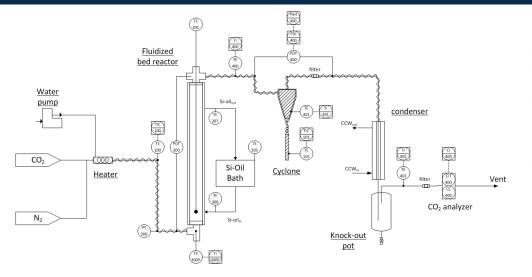
# Large scale production of Hybrid MOF-Based Sorbent

### **Scale-up from 20 mg up to 1Kg**





# CO<sub>2</sub> Adsorption in Visual Fluidized-Bed Reactor



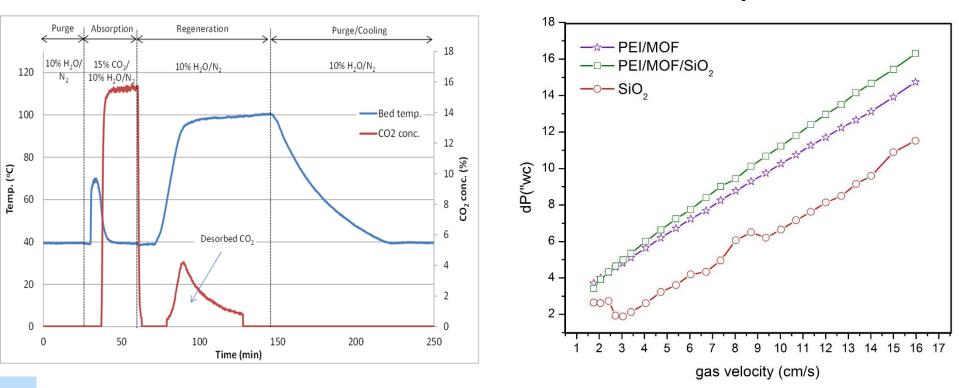


- Verify (visually) the fluidizability of CO<sub>2</sub> capture sorbents
- Operate with realistic process conditions
- Measure  $\triangle P$  and temperature gradients
- Test optimal fluidization conditions
- Fluidized bed reactor allows sorbents to be tested under high water content up to 95% in gas stream
- Bed temperature can be as high as 120 °C
- Pressure drop across the bed is measured over the course of experiment

# Fluidizability Testing & CO<sub>2</sub> breakthrough

### **Breakthrough experiment for PEI/MOF/SiO<sub>2</sub>**

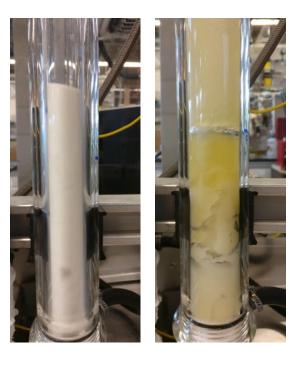
### **Pressure drop measurement**



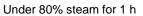
# Fluidizablity Testing Under Humid Stripping Gas

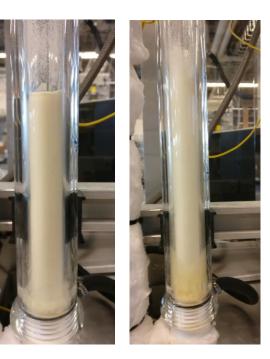
PEI/SiO<sub>2</sub>





Initial

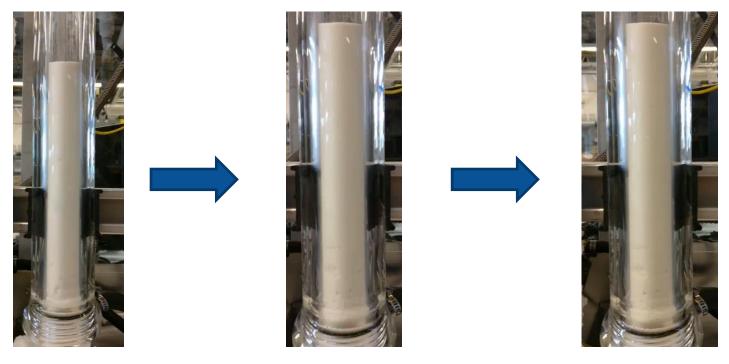




Initial

Under 80% steam for 1 h

# **Fluidizability Testing Under Humid Stripping Gas**



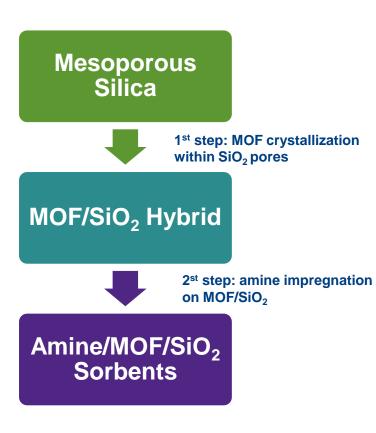
Initial

Regen. under 80% steam for 1 h

Subsequent abs.

**RTI's novel CO<sub>2</sub> fluidized solid sorbents exhibit better performance and long term stability in a fluidized configuration** 

# Hybrid MOF-Based CO<sub>2</sub> Adsorbents



- We developed a very elegant, novel and environmentally friendly way of growing MOF inside the pores of silica that could be extended to other mesoporous supports.
- > We have shown high  $CO_2$  capacity ( $\geq 12$  wt.%) coupled with:
- Excellent MOF dispersion and homogeneity
- Good water and air stability
- Good chemical and thermal stability
- Enhanced attrition resistance
- Excellent fluidizability and solids handling capability
- We are in the process of further testing these hybrid MOF-based CO<sub>2</sub> adsorbents to unveil most of their potential as solid sorbent for CO<sub>2</sub> capture.

# Acknowledgments

➢ Financial support provided by DOE NETL under DE-FE0026432



- DOE Project Manager: Steve Mascaro
- State of North Carolina for Cost Share

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# Thankyou